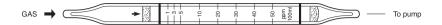
n-VALERIC ACID



1. PERFORMANCE

7) Colour change

1) Measuring range 3-70 ppmNumber of pump strokes $1 (100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and n-Valeric acid concentration is determined by using a conversion chart at

1 pump stroke.∴ Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CH₃ (CH₂) ₃CO₂H + N₂OH → CH₃ (CH₂) ₃COON₂ + H₂O

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	"	The maximum end point of stained layer is indiscernible.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"

