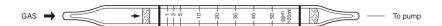
METHACRYLIC ACID



1. PERFORMANCE

7) Colour change

1) Measuring range 1-50 ppmNumber of pump strokes $1(100 \text{m} \ell)$

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years 5) Operating temperature : $15 \sim 25 \,^{\circ}\text{C}$

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Methacrylic acid concentration is determined by using a conversion chart

at 1 pump stroke.

∴ Pale pink → Yellow

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

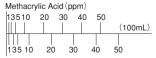
By reacting with alkali, PH indicator is discoloured. $CH_2 = C(CH_3) COOH + NaOH \rightarrow CH_2 = C(CH_3) COONa + H_2O$

4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	//	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"



No.216S tube reading (ppm)